

Atoms	Rank them from Largest to Smallest		
Li, F, C			
Li, Na, K			
Ge, P, O			
C, N, Si			
Al, Cl, Br			
Ba, Mg, Sr			
Atoms	Most Reactive to Least Reactive		
K, Ca, Mg			
F, Cl, S			

Draw in arrows and describe trends in atomic size and reactivity for both metals and non-metals on the periodic table.

Explain trends in atomic size in terms of nuclear charge and electron shielding